## IN THE CLAIMS

- 1. (currently amended) A method of producing a metal mesoporphyrin halide compound comprising: isolating a mesoporphyrin formate; and converting the mesoporphyrin formate to a metal mesoporphyrin halide compound.
- 2. (currently amended) The method of claim 1, wherein the mesoporphyrin formate is converted directly to a the metal mesoporphyrin halide compound.
- 3. (currently amended) The method of claim 1, wherein the mesoporphyrin formate is first converted to mesoporphyrin dihydrochloride and the mesoporphyrin dihydrochloride is converted to the metal mesoporphyrin halide compound.
- 4. (currently amended) The method of claim 3, wherein the mesoporphyrin dihydrochloride is reacted with <u>an</u> insert <u>metals</u> metal to form the metal mesoporphyrin halide <u>compound</u>.
- 5. (previously presented) The method of claim 3, further comprising purifying the mesoporphyrin formate in the presence of a metal scavenger.
- 6. (previously presented) The method of claim 5, wherein the metal scavenger includes Si-thiol.
- 7. (previously presented) The method of claim 4, further comprising catalytically hydrogenating hemin in the presence of an acid to form the mesoporphyrin formate.
- 8. (previously presented) The method of claim 7, wherein the step of catalytically hydrogenating the hemin occurs in two steps.
- 9. (previously presented) The method of claim 8, further comprising heating a mixture of hemin and a hydrogenation catalyst under pressure at a first temperature for

a first period of time and subjecting the mixture to a second temperature under pressure for a second period of time.

- 10. (previously presented) The method of claim 9, wherein the first temperature is higher than the second temperature.
- 11. (currently amended) The method of claim 1, wherein <u>the</u> metal mesoporphyrin <u>halide compound</u> is a <u>tin metal</u> mesoporphyrin halide.
- 12. (currently amended) The process method of claim 10, further comprising: a) subjecting a reaction mixture of hemin and a hydrogenation catalyst in an acid to hydrogen pressure of about 30-65 psi and then raising the temperature to about 85-95° C and maintaining the temperature within that range for a period of about 1-3 hours; b) subjecting the reaction mixture to a further hydrogen pressure of about 30-65 psi at a temperature range of about 45-50° C for a period of about 24-48 hours; and c) recovering the formate salt of mesoporphyrin IX formate from the reaction mixture by precipitation of the mixture with a solvent.
- 13. (currently amended) The process method of claim 7, wherein the acid is formic acid.
- 14. (currently amended) The <u>process method</u> of claim 12, wherein the solvent is an ether.
- 15. (currently amended) The process method of claim 14, wherein the solvent is methyl tert-butyl ether.
- 16. (currently amended) The process method of claim 15, wherein the hydrogenation catalyst is palladium on carbon.

- 17. (currently amended) The method of claim 1, wherein the quantity of metal mesoporphyrin halide compound formed by a single performance of the process method exceeds 0.1 kg.
- 18. (currently amended) The method of claim 1, further comprising purifying the metal mesoporphyrin halide compound, including: a) dissolving the metal mesoporphyrin halide compound in an aqueous basic solution to obtain a dissolved metal mesoporphyrin halide compound; b) treating said dissolved metal mesoporphyrin halide compound; c) adding said treated metal mesoporphyrin halide compound to a first aqueous acid solution to obtain a precipitated metal mesoporphyrin halide compound; d) triturating said precipitated metal mesoporphyrin halide in a second aqueous acid solution at elevated temperature to obtain a substantially pure metal mesoporphyrin halide compound; and e) drying said substantially pure metal mesoporphyrin halide compound.
- 19. (currently amended) The process method of claim 18, wherein the metal mesoporphyrin halide is mesoporphyrin IX chloride dihydrochloride.

## 20.-24. (canceled).

- 25. (new) The method of claim 19, further comprising reacting the mesoporphyrin IX dihydrochloride with tin to form tin V mesoporphyrin IX dichloride.
- 26. (new) The method of claim 1, wherein the mesoporphyrin formate is isolated in substantially pure, solid form.
- 27. (new) The method of claim 26, wherein the mesoporphyrin compound is mesoporphyrin IX dihydrochloride.
- 28. (new) The method of claim 27, further comprising reacting the mesoporphyrin IX dihydrochloride with tin to form tin (IV) mesoporphyrin IX dichloride.
- 29. (new) The method of claim 28, wherein the quantity of tin (IV) mesoporphyrin dichloride formed by a single performance of the method exceeds multiple kilograms.

- 30. (new) The method of claim 28, further comprising purifying the mesoporphyrin formate with a metal scavenger.
- 31. (new) The method of claim 20, wherein the metal scavenger includes a silica bound metal scavenger.
- 32. (new) The method of claim 28, further comprising purifying the tin (IV) mesoporphyrin IX dichloride includes trituration in hot acid at an elevated temperature.
  - 33. (new) The method of claim 32, wherein the acid is HCl.
- 34. (new) The method of claim 11, wherein the metal mesoporphyrin halide is mesoporphyrin IX dihydrochloride.
- 35. (new) The method of claim 18, wherein the metal mesoporphyrin compound is a metal mesoporphyrin halide.
- 36. (new) The method of claim 4, wherein the mesoporphyrin dihydrochloride is reacted with the insert metal in the absence of any acetate ions.
- 37. (new) The method of claim 28, wherein the mesoporphyrin dihydrochloride is reacted with the tin insert metal in the absence of any acetate ions.
- 38. (new) The method of claim 17, wherein the quantity of metal mesoporphyrin compound formed by a single performance of the method exceeds 1 kg.
- 39. (new) The method of claim 18, wherein the quantity of metal mesoporphyrin compound formed by a single performance of the method exceeds 1 kg.
- 40. (new) The method of claim 28, wherein the quantity of tin (IV) mesoporphyrin IX dichloride formed by a single performance of the method exceeds 1 kg.
- 41. (new) Tin (IV) mesoporphyrin IX dichloride produced by a single performance of the method of claim 28.